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clitics are grammatical elements that are treated as independent words in syntax but form a phonological unit with the word that precedes or follows it this volume brings together the facts about clitics in the slavic languages where they have become a focal points of recent research the authors draw relevant generalizations across the slavic languages and highlight the importance of these phenomena for linguistic theory

prior to 1862 when the department of agriculture was established the report on agriculture was prepared and published by the commissioner of patents and forms volume or part of volume of his annual reports the first being that of 1840 cf checklist of public documents washington 1895 p 148

czech a clear case of a language having a standard and a strong central vernacular with intensive shifting between them offers many points of general interest to sociolinguists this volume is divided in 5 chapters and opens with a general discussion of language varieties the two central language formations in czech gives a summary description of the czech central vernacular this is followed by a chapter on the origin and opposition of standard and common czech and in the next chapter code switching between standard and common czech is discussed the concluding chapter presents starting points for a theoretical description of a national language with intralingual variation and a preliminary formulation of perspectives on the stratification of czech

keep up to date with recent research and practice in suicide this book is based on the triple i in suicidology international conferences which are organised annually by the slovene centre for suicide research in memory of the late prof andrej marušič with the aim of promoting intuition imagination and innovation in the research and prevention of suicide and suicidal behaviour the carefully selected chapters provide food for thought to practitioners researchers students and all those who come into contact with the tragedy of suicide with the hope of stimulating new ideas and interventions in the difficult fight against suicidal behaviours in four parts the internationally renowned team of authors summarise the achievements of suicidology so far both in quantitative and qualitative research present effective interventions in suicide prevention including for youths and older people and knowledge gained in bereavement and postvention studies such as in different cultures and those bereaved by suicide and highlight future directions for suicide research and prevention the volume is thus a useful resource for all those interested in keeping up to date with recent research and practice in suicide

this book deals with the works of ivan cankar the greatest slovenian writer focusing on his relation to existential social and moral reality as reflected in individuals and in society at large the method of literary analysis shows a surprising harmony between personal confessions and a rich symbolism that reveals the writer s unconditional belief in the power of conscience strong conviction of the sense of victims and the longing for the triumph of love and justice a holistic interpretation yields the conclusion that most of cankar s works are confessions that purport to be true to life his inclination to self disclosure in dreams alongside the objective disclosure of imperceptible reality indicates that expressive language and a lyrical style are of vital importance to him

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in the world of magazines no recognition is more highly coveted than an ellie the national magazine award presented by the american society of magazine editors to the best of the american magazines the awards are the magazine equivalents to the pulitzer prizes of the newspaper industry each year hundreds of editors in chief journalism professors and art directors winnow more than a thousand submissions to about seventy five nominees in categories such as reporting feature writing profiles public interest essays reviews and criticism interest in the nominees is keen and this collection will allow people both in the magazine world and beyond to find in one place read and admire the year s best it is a wonderful browsable volume of interest to writers and readers who appreciate magazine writing and journalism at its highest level

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engaging with pre feminist and male authored crime literature resisting invisibility offers a comparative reading of women s bodies as represented in spanish crime literature from the mid nineteenth century to the present utilizing the twin concepts of visibility and invisibility the book establishes a genealogy of differing viewpoints regarding women s positions in these narratives before and after the birth of the modern spanish female detective

this examination of the politics of female visibility expands our understanding of the aesthetic regimes that have governed the female body from the early phases of the genre's evolution while most scholars understand the feminization of the crime genre as a response to second wave feminism resisting invisibility demonstrates that even in the earliest representations of delinquent women the politics surrounding the female body are problematized and are more complex than previously conceptualized drawing on gender and queer studies resisting invisibility investigates the gendering of crime fiction forcing us to reconsider the literary history of female visibility and prompting us to establish an alternative genealogy for spanish crime literature

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Unveiling the Enigma of C₆H₅OH: Phenol's Chemistry, Applications, and Safety

The chemical formula C₆H₅OH might seem cryptic at first glance, but it represents a compound with a surprisingly diverse range of applications and a significant impact on our daily lives. This compound, known as phenol, is a fascinating example of how seemingly simple chemical structures can exhibit complex properties and play crucial roles in various industries. Understanding phenol requires delving into its chemical structure, its reactivity, its uses, and, critically, its safety precautions. This article aims to provide a comprehensive overview of C₆H₅OH, catering to readers seeking in-depth knowledge and practical insights.

1. Understanding the Structure and Properties of Phenol: Phenol's structure is crucial to understanding its behaviour. It consists of a benzene ring (C₆H₅) directly bonded to a hydroxyl group (-OH). This seemingly simple addition of the hydroxyl group to the aromatic ring dramatically alters the properties of benzene, transforming it from a relatively inert hydrocarbon into a reactive and versatile compound. The hydroxyl group's presence introduces polarity to the molecule, making phenol slightly soluble in water. This solubility, however, is limited compared to simpler alcohols. The hydroxyl group also acts as a weak acid, capable of donating a proton (H⁺) in aqueous solutions, resulting in a slightly acidic pH. This acidity is a key factor in many of phenol's applications and reactions. The benzene ring, with its delocalized electrons, contributes to phenol's stability and allows it to participate in various electrophilic aromatic substitution reactions.

2. Chemical Reactivity of Phenol: The hydroxyl group's influence on the benzene ring's reactivity is profound. The electron-donating nature of the -OH group activates the ring, making it more susceptible to electrophilic attack. This explains phenol's tendency to undergo various substitution reactions more readily than benzene itself. Common reactions include:

- Nitration: The reaction with nitric acid produces nitrophenols, crucial intermediates in the synthesis of various dyes and explosives. Careful control of reaction conditions is essential, as over-nitration can occur.
- Halogenation: Reaction with halogens like bromine leads to the formation of halophenols, used in the production of disinfectants and pesticides.
- Sulfonation: Reaction with sulfuric acid produces phenolsulfonic acids, important intermediates in the synthesis of dyes and pharmaceuticals.

Understanding these reactions is crucial for anyone working with phenol in a laboratory or industrial setting.

3. Applications of Phenol: Phenol's versatility has led to its widespread use in diverse industries:

- Disinfectants and Antiseptics: Phenol was historically the first antiseptic used in surgery, though its use is now limited due to its toxicity. Derivatives of phenol, like cresols and hexachlorophene, are still employed in some disinfectants and antiseptics.
- Resin Production: Phenol is a key component in the manufacture of phenolic resins, a type of thermosetting polymer used extensively in adhesives, coatings, and laminates (e.g., plywood). Bakelite, the first fully

synthetic plastic, is a famous example of a phenolic resin. Pharmaceutical Industry: Phenol and its derivatives serve as building blocks for various pharmaceuticals, including aspirin and many other drugs. Agriculture: Certain phenol derivatives are used as herbicides and pesticides to control weeds and pests. 4. Safety Considerations and Handling of Phenol: Phenol is a corrosive and toxic substance. Skin contact can cause burns and irritation, while ingestion can be fatal. Inhalation of phenol vapors can also lead to respiratory problems. Therefore, handling phenol requires strict adherence to safety protocols: Personal Protective Equipment (PPE): Always wear appropriate PPE, including gloves, eye protection, and lab coats, when handling phenol. Ventilation: Work in well-ventilated areas to minimize exposure to phenol vapors. Spill Response: Have a plan in place to handle spills safely, using appropriate absorbent materials and neutralizing agents. Disposal: Dispose of phenol waste according to local regulations. 5. Conclusion: C_6H_5OH , or phenol, is a fascinating compound with a rich history and a wide range of applications. Understanding its chemical structure, reactivity, and potential hazards is crucial for safe and effective utilization. From its role in the creation of everyday materials like plastics to its historical significance in medicine, phenol continues to hold a prominent place in chemistry and industry. Always prioritize safety when handling this versatile yet potentially hazardous compound. FAQs: 1. Is phenol biodegradable? Phenol is considered biodegradable, but the rate depends on environmental conditions. Microbial degradation is the primary mechanism. 2. What is the difference between phenol and benzene? Phenol is benzene with a hydroxyl (-OH) group attached. This seemingly small change significantly alters its properties, making phenol reactive and acidic, unlike the relatively inert benzene. 3. What are the common symptoms of phenol poisoning? Symptoms can range from mild skin irritation to severe burns, respiratory distress, and cardiovascular collapse depending on the route and degree of exposure. Immediate medical attention is vital. 4. Are there any safer alternatives to phenol as a disinfectant? Yes, several alternatives exist, including alcohols (ethanol, isopropanol), quaternary ammonium compounds, and hydrogen peroxide. The choice of alternative depends on the specific application. 5. What is the industrial process for phenol production? The most common industrial process is the cumene process, which involves the oxidation of cumene (isopropylbenzene) to produce phenol and acetone as coproducts.

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